# C18-AA/A/CH/E1/FW/C/CM/EC/EE/ M/MET/MNG/PKG/TT-204

# 6204

## BOARD DIPLOMA EXAMINATION, (C-18)

MARCH / APRIL - 2019

### COMMON - II SEMESTER EXAMINATION

### **ENGINEERING CHEMISTRY &** ENVIRONMENTAL STUDIES

Time : 2 Hours]

[Total Marks: 40

PART - A

08×01=08

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Instructions:

- (1) Answer ALL questions.
- (2) Each question carries ONE marks.
- Define flux.

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- What are the different types of plastics?
- 3 Define Galvanic cell.
- What is the chemical formula of rust?
- What is electrochemical series?
- What are the different types of batteries?
- Define primary pollutant.
- Define water pollution.

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Contd...

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PART - B

04×03=12

Instructions:

- (1) Answer any FOUR questions.
- Each question carries THREE marks.

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Write any six characteristics of metals.

OR

- Write any three significances of electrochemical series.
- (a) Write a method of preparation and two uses of polystyrene.

OR

- (a) Write any three differences between electrolytic cell and galvanic cell. http://www.sbtetonline.com

- Write the electrode reaction of lead storage battery during discharging.
- Write a short note on green house effect.

Write a note on ozone layer depletion.

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Classify the air pollutants based on origin and give examples.

OR

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PART - C

04205-20

Instructions :

- (1) Answer any FOUR questions.
- (2) Each question carries FIVE marks.
- 13 (a) Explain the formation of composition cell and stress cell.

OR

- (b) Describe the construction and working principle of Leclanche cell.
- 14 (a) Calculate the mass and volume of oxygen required for the complete combustion of 15 g of ethane (C<sub>2</sub>H<sub>6</sub>)

OR

- (b) Explain the causes of water pollution.
- 15 (a) Explain the construction and functioning of Galvanic cell-

OR

- (b) Explain working principle of H2O2 Fuel cell.
- 16 (a) Explain the control methods of water pollution.

OR

(b) Explain the control methods of air pollution.

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